

SUPPLEMENTAL HW 11

1 & 2)

$$n = 142 = DP$$

SO MOST CHAINS HAVE 142 MONOMER UNITS

$$\text{MONOMER WEIGHT OF MONOMER} = 28 \frac{\text{g}}{\text{mol}}$$

$$142 \cdot 28 \frac{\text{g}}{\text{mol}} = \boxed{3,976 \frac{\text{g}}{\text{mol}} = M_n}$$

3) "LIVING" MEANS CHAINS DON'T CONNECT

$$\frac{10 \text{ mmol INITIATOR} \mid 1 \text{ mol} \mid 2 \text{ mol POLYMER CHAIN}}{1000 \text{ mmol} \mid 1 \text{ mol INITIATOR}} = 0.02 \text{ mol POLYMERS}$$

$$MW \text{ MONOMER} = 200 \frac{\text{g}}{\text{mol}}$$

$$\frac{1.2 \text{ kg MONOMER} \mid 1000 \text{ g} \mid 1 \text{ mol} \mid 0.5 \text{ mol MONOMER REACTED}}{1 \text{ kg} \mid 200 \text{ g} \mid 1 \text{ mol MONOMER IN BEGINNING}} = 3 \text{ mol MONOMER REACTED}$$

$$n = \frac{3 \text{ mol MONOMER IN POLYMER}}{0.02 \text{ mol POLYMER}} = \boxed{150}$$